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***N*-Methyl-1,2-diphenylcyclobuteno-[3,4-*a*]naphthalene-2,3-dicarboximide**

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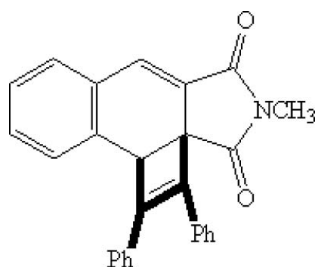
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.058; wR factor = 0.123; data-to-parameter ratio = 13.7.

The title compound, $\text{C}_{27}\text{H}_{19}\text{NO}_2$, is the product of a photochemical reaction between *N*-methyl-naphthalene-2,3-dicarboximide and diphenylacetylene. The cyclobutene ring in the molecule is almost perpendicular to the plane of the dihydronaphthalimide system; the interplanar angle between the cyclobutene ring and the plane through the two six-membered rings of the naphthalimide system is 70.66 (9)°. The interplanar angle between the two phenyl substituents on the cyclobutene ring is 42.47 (8)°.

Related literature

For information on the photochemistry of naphthalene-dicarboximides with alkynes see Liu *et al.* (2006) and for related structures see Liu *et al.* (2003, 2006).



Experimental

Crystal data

$\text{C}_{27}\text{H}_{19}\text{NO}_2$
 $M_r = 389.43$
 Monoclinic, $P2_1/c$
 $a = 10.763$ (2) Å
 $b = 7.9925$ (15) Å
 $c = 23.438$ (5) Å
 $\beta = 97.326$ (3)°

$V = 1999.8$ (7) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 298$ (2) K
 $0.28 \times 0.22 \times 0.08$ mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 Absorption correction: none
 10516 measured reflections

3715 independent reflections
 2456 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.062$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.058$
 $wR(F^2) = 0.123$
 $S = 1.01$
 3715 reflections

272 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.17$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.15$ e Å⁻³

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2001); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SJ2314).

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supplementary materials

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N-Methyl-1,2-diphenylcyclobuteno[3,4-*a*]naphthalene-2,3-dicarboximide

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Comment

In the studies of photochemistry of naphthalenedicarboximides with alkynes, it has been found that the photoinduced cycloadditions of *N*-methyl-1,8-naphthalenedicarboximides (NI18) with alkynes followed the pathway of NI18 with alkenes (Liu *et al.*, 2006), but that is not the case for *N*-methylnaphthalene-2,3-dicarboximide (NI23). UV irradiation of NI23 in the presence of diphenylacetylene in benzene solution resulted in the novel adduct (I) and its structure is reported here (Figure 1).

The bond lengths and angles in (I) show normal values except for the geometry of the cyclobutene ring (Table 1). The C20—C21 bond length of 1.345 (3) Å is nearly the same as that in 4-methyl-1,2-diphenylbenzo[de]cyclobut[*i*]isoquinoline-3,5(2*H*)-dione(1.346 (3) Å) (Liu *et al.*, 2006), but is longer than those of 4,5,9,10-tetrahydro-4-methyl-2-phenyl-9,10-epoxy-3*H*,10*aH*-cyclobuta[benzo[2,3,4-*de*]-isoquinoline-3,5-dione (1.324 (2) Å) (Liu *et al.*, 2003), Such elongation is considered to be caused by the steric effect of the bulky phenyl substituents on C20 and C21, respectively. The cyclobutene ring in the molecule is almost perpendicular to the plane of dihydronaphthalimide, the interplanar angle between the cyclobutene ring and the plane through the two six membered rings of the naphthalimide is 70.66 (9). The interplanar angle between the two benzene substituents on the cyclobutene ring is 42.47 (8).

Experimental

The title compound, (I), was prepared by irradiation of a benzene (150 ml) solution of *N*-methylnaphthalene-2,3-dicarboximide (633 mg, 3.0 mmol) with diphenylacetylene (1.068 g, 6 mmol) with UV light of wavelength longer than 300 nm purged with nitrogen, and isolated by flash column chromatography on silica gel, eluting with petroleum-ethyl acetate (4:1 v/v). Single crystals (m.p. 505–506 K) suitable for X-ray diffraction were obtained by recrystallization from ethanol-ethyl acetate solution (1:95 v/v).

Refinement

Hydrogen atoms were placed in geometrically idealized positions and refined using a riding model with $d(\text{C—H}) = 0.93$ Å, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for aromatic 0.98 Å, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for CH and 0.96 Å, $U_{\text{iso}} = 1.5U_{\text{eq}}(\text{C})$ for CH₃ atoms.

Figures

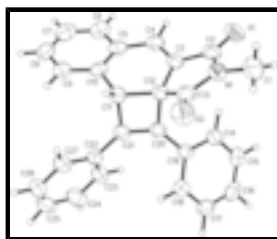


Fig. 1. The structure of (I) with the atom-numbering scheme and ellipsoids drawn at the 30% probability level.



Fig. 2. The formation of (I).

N-Methyl-1,2-diphenylcyclobuteno[3,4-*a*]naphthalene-2,3-dicarboximide

Crystal data

$C_{27}H_{19}NO_2$	$F_{000} = 816$
$M_r = 389.43$	$D_x = 1.293 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Melting point: 505 K
Hall symbol: -P 2ybc	Mo $K\alpha$ radiation
$a = 10.763 (2) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$b = 7.9925 (15) \text{ \AA}$	Cell parameters from 1602 reflections
$c = 23.438 (5) \text{ \AA}$	$\theta = 2.4\text{--}20.9^\circ$
$\beta = 97.326 (3)^\circ$	$\mu = 0.08 \text{ mm}^{-1}$
$V = 1999.8 (7) \text{ \AA}^3$	$T = 298 (2) \text{ K}$
$Z = 4$	Plate, colourless
	$0.28 \times 0.22 \times 0.08 \text{ mm}$

Data collection

Bruker SMART APEX CCD area-detector diffractometer	2456 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.062$
Monochromator: graphite	$\theta_{\text{max}} = 25.5^\circ$
$T = 298(2) \text{ K}$	$\theta_{\text{min}} = 1.9^\circ$
φ and ω scans	$h = -11 \rightarrow 13$
Absorption correction: none	$k = -9 \rightarrow 7$
10516 measured reflections	$l = -28 \rightarrow 24$
3715 independent reflections	

Refinement

Refinement on F^2	H-atom parameters constrained
Least-squares matrix: full	$w = 1/[\sigma^2(F_o^2) + (0.0426P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.058$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.123$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.01$	$\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$
3715 reflections	$\Delta\rho_{\text{min}} = -0.15 \text{ e \AA}^{-3}$
272 parameters	Extinction correction: none
Primary atom site location: structure-invariant direct methods	
Secondary atom site location: difference Fourier map	
Hydrogen site location: inferred from neighbouring sites	

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.5270 (3)	0.7812 (4)	0.55850 (12)	0.0659 (9)
H1A	0.5237	0.7875	0.5174	0.099*
H1B	0.6028	0.7257	0.5743	0.099*
H1C	0.5256	0.8921	0.5741	0.099*
C2	0.4147 (2)	0.5119 (3)	0.57395 (9)	0.0415 (6)
C3	0.2884 (2)	0.4690 (3)	0.58640 (9)	0.0362 (6)
C4	0.2254 (2)	0.3278 (3)	0.57293 (9)	0.0417 (6)
H4	0.2692	0.2324	0.5647	0.050*
C5	0.0900 (2)	0.3192 (3)	0.57082 (9)	0.0406 (6)
C6	0.0276 (3)	0.1680 (3)	0.55931 (10)	0.0528 (7)
H6	0.0738	0.0705	0.5568	0.063*
C7	-0.1008 (3)	0.1601 (4)	0.55158 (11)	0.0608 (8)
H7	-0.1408	0.0576	0.5448	0.073*
C8	-0.1705 (3)	0.3042 (4)	0.55384 (11)	0.0577 (8)
H8	-0.2576	0.3000	0.5480	0.069*
C9	-0.1089 (2)	0.4554 (4)	0.56487 (10)	0.0501 (7)
H9	-0.1560	0.5526	0.5658	0.060*
C10	0.0197 (2)	0.4662 (3)	0.57446 (9)	0.0389 (6)
C11	0.0817 (2)	0.6310 (3)	0.59191 (10)	0.0384 (6)
H11	0.0499	0.7255	0.5677	0.046*
C12	0.2275 (2)	0.6241 (3)	0.60469 (9)	0.0361 (6)
C13	0.3101 (2)	0.7618 (3)	0.58514 (10)	0.0428 (6)
C14	0.4107 (2)	0.5646 (3)	0.72463 (10)	0.0410 (6)
H14	0.4304	0.5042	0.6931	0.049*
C15	0.4926 (2)	0.5629 (3)	0.77487 (11)	0.0477 (7)
H15	0.5656	0.4997	0.7773	0.057*
C16	0.4659 (2)	0.6550 (3)	0.82139 (11)	0.0506 (7)
H16	0.5206	0.6538	0.8555	0.061*
C17	0.3578 (2)	0.7489 (3)	0.81730 (11)	0.0496 (7)
H17	0.3406	0.8131	0.8484	0.060*
C18	0.2754 (2)	0.7483 (3)	0.76758 (10)	0.0435 (6)
H18	0.2023	0.8114	0.7655	0.052*
C19	0.2998 (2)	0.6539 (3)	0.71998 (9)	0.0342 (6)
C20	0.2084 (2)	0.6472 (3)	0.66827 (10)	0.0360 (6)

supplementary materials

C21	0.0834 (2)	0.6619 (3)	0.65621 (10)	0.0373 (6)
C22	-0.0208 (2)	0.6824 (3)	0.69000 (10)	0.0386 (6)
C23	-0.0146 (2)	0.6137 (3)	0.74466 (11)	0.0475 (7)
H23	0.0551	0.5512	0.7594	0.057*
C24	-0.1108 (3)	0.6373 (3)	0.77724 (12)	0.0585 (8)
H24	-0.1054	0.5916	0.8140	0.070*
C25	-0.2141 (3)	0.7276 (4)	0.75599 (15)	0.0628 (9)
H25	-0.2781	0.7451	0.7785	0.075*
C26	-0.2234 (3)	0.7922 (4)	0.70151 (15)	0.0635 (9)
H26	-0.2948	0.8510	0.6866	0.076*
C27	-0.1271 (2)	0.7703 (3)	0.66871 (12)	0.0521 (7)
H27	-0.1338	0.8152	0.6318	0.062*
N1	0.41972 (19)	0.6876 (3)	0.57308 (8)	0.0442 (5)
O1	0.50087 (17)	0.4229 (2)	0.56464 (7)	0.0553 (5)
O2	0.28991 (17)	0.9106 (2)	0.58275 (8)	0.0599 (5)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.056 (2)	0.076 (2)	0.068 (2)	-0.0136 (16)	0.0180 (15)	-0.0095 (16)
C2	0.0456 (17)	0.0540 (18)	0.0244 (13)	0.0114 (13)	0.0026 (11)	0.0012 (12)
C3	0.0397 (15)	0.0406 (15)	0.0282 (13)	0.0099 (12)	0.0038 (11)	0.0018 (11)
C4	0.0517 (18)	0.0376 (15)	0.0352 (14)	0.0127 (13)	0.0033 (12)	0.0013 (12)
C5	0.0518 (17)	0.0427 (16)	0.0270 (13)	0.0008 (13)	0.0044 (11)	0.0015 (12)
C6	0.066 (2)	0.0465 (17)	0.0441 (17)	-0.0019 (15)	0.0011 (14)	-0.0001 (13)
C7	0.074 (2)	0.062 (2)	0.0461 (17)	-0.0197 (17)	0.0068 (15)	-0.0055 (15)
C8	0.0507 (18)	0.082 (2)	0.0396 (16)	-0.0136 (17)	0.0035 (13)	-0.0123 (16)
C9	0.0482 (18)	0.066 (2)	0.0352 (15)	0.0011 (14)	0.0032 (12)	-0.0057 (14)
C10	0.0445 (16)	0.0473 (16)	0.0247 (13)	0.0014 (13)	0.0037 (11)	-0.0009 (11)
C11	0.0418 (16)	0.0352 (15)	0.0369 (14)	0.0096 (11)	0.0000 (11)	0.0009 (11)
C12	0.0393 (15)	0.0360 (14)	0.0333 (14)	0.0066 (11)	0.0064 (11)	0.0002 (11)
C13	0.0472 (17)	0.0475 (17)	0.0331 (15)	0.0040 (14)	0.0025 (12)	-0.0012 (13)
C14	0.0384 (16)	0.0432 (15)	0.0424 (15)	0.0003 (12)	0.0084 (12)	-0.0016 (12)
C15	0.0352 (16)	0.0543 (18)	0.0522 (17)	-0.0018 (13)	-0.0001 (13)	0.0018 (14)
C16	0.0435 (17)	0.0651 (19)	0.0406 (16)	-0.0135 (14)	-0.0051 (12)	0.0041 (14)
C17	0.0517 (18)	0.0574 (18)	0.0405 (16)	-0.0093 (14)	0.0084 (13)	-0.0111 (13)
C18	0.0406 (16)	0.0454 (15)	0.0451 (16)	0.0014 (12)	0.0082 (12)	-0.0047 (13)
C19	0.0350 (14)	0.0331 (14)	0.0354 (14)	-0.0008 (11)	0.0077 (11)	0.0003 (11)
C20	0.0385 (15)	0.0330 (14)	0.0367 (14)	0.0056 (11)	0.0050 (11)	-0.0019 (11)
C21	0.0411 (16)	0.0328 (14)	0.0378 (14)	0.0070 (11)	0.0039 (11)	-0.0013 (11)
C22	0.0358 (15)	0.0355 (14)	0.0439 (16)	0.0039 (11)	0.0030 (11)	-0.0107 (12)
C23	0.0424 (17)	0.0458 (17)	0.0556 (18)	0.0050 (12)	0.0110 (13)	0.0006 (14)
C24	0.063 (2)	0.0551 (19)	0.0621 (19)	-0.0040 (15)	0.0253 (16)	-0.0019 (15)
C25	0.051 (2)	0.058 (2)	0.086 (3)	-0.0054 (16)	0.0312 (17)	-0.0236 (18)
C26	0.0395 (18)	0.067 (2)	0.083 (2)	0.0123 (14)	0.0045 (16)	-0.0260 (18)
C27	0.0470 (18)	0.0579 (18)	0.0501 (17)	0.0131 (14)	0.0016 (13)	-0.0109 (14)
N1	0.0416 (13)	0.0496 (14)	0.0422 (13)	-0.0013 (11)	0.0084 (10)	-0.0006 (10)
O1	0.0509 (12)	0.0680 (13)	0.0486 (12)	0.0213 (10)	0.0125 (9)	0.0012 (9)

O2 0.0661 (14) 0.0384 (11) 0.0755 (14) 0.0037 (10) 0.0095 (10) 0.0017 (10)

Geometric parameters (Å, °)

C1—N1	1.453 (3)	C13—O2	1.208 (3)
C1—H1A	0.9600	C13—N1	1.382 (3)
C1—H1B	0.9600	C14—C15	1.378 (3)
C1—H1C	0.9600	C14—C19	1.383 (3)
C2—O1	1.210 (3)	C14—H14	0.9300
C2—N1	1.406 (3)	C15—C16	1.376 (3)
C2—C3	1.467 (3)	C15—H15	0.9300
C3—C4	1.333 (3)	C16—C17	1.378 (3)
C3—C12	1.491 (3)	C16—H16	0.9300
C4—C5	1.453 (3)	C17—C18	1.371 (3)
C4—H4	0.9300	C17—H17	0.9300
C5—C6	1.392 (3)	C18—C19	1.399 (3)
C5—C10	1.407 (3)	C18—H18	0.9300
C6—C7	1.372 (4)	C19—C20	1.461 (3)
C6—H6	0.9300	C20—C21	1.345 (3)
C7—C8	1.379 (4)	C21—C22	1.462 (3)
C7—H7	0.9300	C22—C27	1.381 (3)
C8—C9	1.387 (3)	C22—C23	1.388 (3)
C8—H8	0.9300	C23—C24	1.376 (3)
C9—C10	1.376 (3)	C23—H23	0.9300
C9—H9	0.9300	C24—C25	1.365 (4)
C10—C11	1.509 (3)	C24—H24	0.9300
C11—C21	1.525 (3)	C25—C26	1.369 (4)
C11—C12	1.560 (3)	C25—H25	0.9300
C11—H11	0.9800	C26—C27	1.378 (4)
C12—C13	1.522 (3)	C26—H26	0.9300
C12—C20	1.541 (3)	C27—H27	0.9300
N1—C1—H1A	109.5	N1—C13—C12	107.3 (2)
N1—C1—H1B	109.5	C15—C14—C19	121.6 (2)
H1A—C1—H1B	109.5	C15—C14—H14	119.2
N1—C1—H1C	109.5	C19—C14—H14	119.2
H1A—C1—H1C	109.5	C16—C15—C14	119.8 (2)
H1B—C1—H1C	109.5	C16—C15—H15	120.1
O1—C2—N1	123.5 (2)	C14—C15—H15	120.1
O1—C2—C3	130.5 (3)	C15—C16—C17	119.8 (2)
N1—C2—C3	105.9 (2)	C15—C16—H16	120.1
C4—C3—C2	127.4 (2)	C17—C16—H16	120.1
C4—C3—C12	122.9 (2)	C18—C17—C16	120.4 (2)
C2—C3—C12	108.4 (2)	C18—C17—H17	119.8
C3—C4—C5	121.7 (2)	C16—C17—H17	119.8
C3—C4—H4	119.1	C17—C18—C19	120.8 (2)
C5—C4—H4	119.1	C17—C18—H18	119.6
C6—C5—C10	119.1 (3)	C19—C18—H18	119.6
C6—C5—C4	120.3 (2)	C14—C19—C18	117.6 (2)
C10—C5—C4	120.2 (2)	C14—C19—C20	121.9 (2)

supplementary materials

C7—C6—C5	121.2 (3)	C18—C19—C20	120.5 (2)
C7—C6—H6	119.4	C21—C20—C19	136.1 (2)
C5—C6—H6	119.4	C21—C20—C12	93.44 (19)
C6—C7—C8	120.0 (3)	C19—C20—C12	130.4 (2)
C6—C7—H7	120.0	C20—C21—C22	135.4 (2)
C8—C7—H7	120.0	C20—C21—C11	94.53 (19)
C7—C8—C9	119.0 (3)	C22—C21—C11	129.8 (2)
C7—C8—H8	120.5	C27—C22—C23	118.3 (2)
C9—C8—H8	120.5	C27—C22—C21	121.3 (2)
C10—C9—C8	122.2 (3)	C23—C22—C21	120.4 (2)
C10—C9—H9	118.9	C24—C23—C22	120.5 (2)
C8—C9—H9	118.9	C24—C23—H23	119.8
C9—C10—C5	118.4 (2)	C22—C23—H23	119.8
C9—C10—C11	120.0 (2)	C25—C24—C23	120.4 (3)
C5—C10—C11	121.5 (2)	C25—C24—H24	119.8
C10—C11—C21	111.00 (19)	C23—C24—H24	119.8
C10—C11—C12	114.87 (19)	C24—C25—C26	119.9 (3)
C21—C11—C12	86.05 (17)	C24—C25—H25	120.0
C10—C11—H11	114.0	C26—C25—H25	120.0
C21—C11—H11	114.0	C25—C26—C27	120.1 (3)
C12—C11—H11	114.0	C25—C26—H26	120.0
C3—C12—C13	102.7 (2)	C27—C26—H26	120.0
C3—C12—C20	119.68 (19)	C26—C27—C22	120.7 (3)
C13—C12—C20	111.06 (19)	C26—C27—H27	119.6
C3—C12—C11	116.3 (2)	C22—C27—H27	119.6
C13—C12—C11	121.9 (2)	C13—N1—C2	113.0 (2)
C20—C12—C11	85.83 (17)	C13—N1—C1	123.5 (2)
O2—C13—N1	124.5 (2)	C2—N1—C1	123.4 (2)
O2—C13—C12	128.1 (2)		
O1—C2—C3—C4	-23.2 (4)	C16—C17—C18—C19	-0.6 (4)
N1—C2—C3—C4	155.4 (2)	C15—C14—C19—C18	2.4 (3)
O1—C2—C3—C12	170.0 (2)	C15—C14—C19—C20	-175.7 (2)
N1—C2—C3—C12	-11.4 (2)	C17—C18—C19—C14	-1.3 (3)
C2—C3—C4—C5	-160.1 (2)	C17—C18—C19—C20	176.8 (2)
C12—C3—C4—C5	4.9 (3)	C14—C19—C20—C21	150.3 (3)
C3—C4—C5—C6	-176.9 (2)	C18—C19—C20—C21	-27.7 (4)
C3—C4—C5—C10	9.7 (3)	C14—C19—C20—C12	-31.9 (4)
C10—C5—C6—C7	-0.1 (4)	C18—C19—C20—C12	150.1 (2)
C4—C5—C6—C7	-173.6 (2)	C3—C12—C20—C21	-121.1 (2)
C5—C6—C7—C8	1.5 (4)	C13—C12—C20—C21	119.6 (2)
C6—C7—C8—C9	-1.0 (4)	C11—C12—C20—C21	-3.11 (19)
C7—C8—C9—C10	-0.9 (4)	C3—C12—C20—C19	60.5 (3)
C8—C9—C10—C5	2.3 (4)	C13—C12—C20—C19	-58.9 (3)
C8—C9—C10—C11	-174.1 (2)	C11—C12—C20—C19	178.5 (2)
C6—C5—C10—C9	-1.8 (3)	C19—C20—C21—C22	-4.2 (5)
C4—C5—C10—C9	171.7 (2)	C12—C20—C21—C22	177.5 (3)
C6—C5—C10—C11	174.6 (2)	C19—C20—C21—C11	-178.5 (3)
C4—C5—C10—C11	-11.9 (3)	C12—C20—C21—C11	3.18 (19)
C9—C10—C11—C21	81.3 (3)	C10—C11—C21—C20	112.0 (2)

C5—C10—C11—C21	-95.0 (2)	C12—C11—C21—C20	-3.14 (19)
C9—C10—C11—C12	176.8 (2)	C10—C11—C21—C22	-62.8 (3)
C5—C10—C11—C12	0.5 (3)	C12—C11—C21—C22	-177.9 (2)
C4—C3—C12—C13	-151.6 (2)	C20—C21—C22—C27	148.6 (3)
C2—C3—C12—C13	16.0 (2)	C11—C21—C22—C27	-38.8 (4)
C4—C3—C12—C20	84.9 (3)	C20—C21—C22—C23	-31.2 (4)
C2—C3—C12—C20	-107.6 (2)	C11—C21—C22—C23	141.4 (2)
C4—C3—C12—C11	-15.9 (3)	C27—C22—C23—C24	-1.9 (4)
C2—C3—C12—C11	151.61 (19)	C21—C22—C23—C24	177.9 (2)
C10—C11—C12—C3	12.6 (3)	C22—C23—C24—C25	0.6 (4)
C21—C11—C12—C3	123.9 (2)	C23—C24—C25—C26	1.3 (4)
C10—C11—C12—C13	139.1 (2)	C24—C25—C26—C27	-1.8 (4)
C21—C11—C12—C13	-109.5 (2)	C25—C26—C27—C22	0.4 (4)
C10—C11—C12—C20	-108.6 (2)	C23—C22—C27—C26	1.4 (4)
C21—C11—C12—C20	2.74 (16)	C21—C22—C27—C26	-178.4 (2)
C3—C12—C13—O2	168.5 (2)	O2—C13—N1—C2	-174.5 (2)
C20—C12—C13—O2	-62.4 (3)	C12—C13—N1—C2	8.8 (3)
C11—C12—C13—O2	36.1 (4)	O2—C13—N1—C1	3.0 (4)
C3—C12—C13—N1	-15.0 (2)	C12—C13—N1—C1	-173.8 (2)
C20—C12—C13—N1	114.1 (2)	O1—C2—N1—C13	-179.9 (2)
C11—C12—C13—N1	-147.4 (2)	C3—C2—N1—C13	1.5 (3)
C19—C14—C15—C16	-1.6 (4)	O1—C2—N1—C1	2.7 (4)
C14—C15—C16—C17	-0.3 (4)	C3—C2—N1—C1	-175.9 (2)
C15—C16—C17—C18	1.4 (4)		

Fig. 1

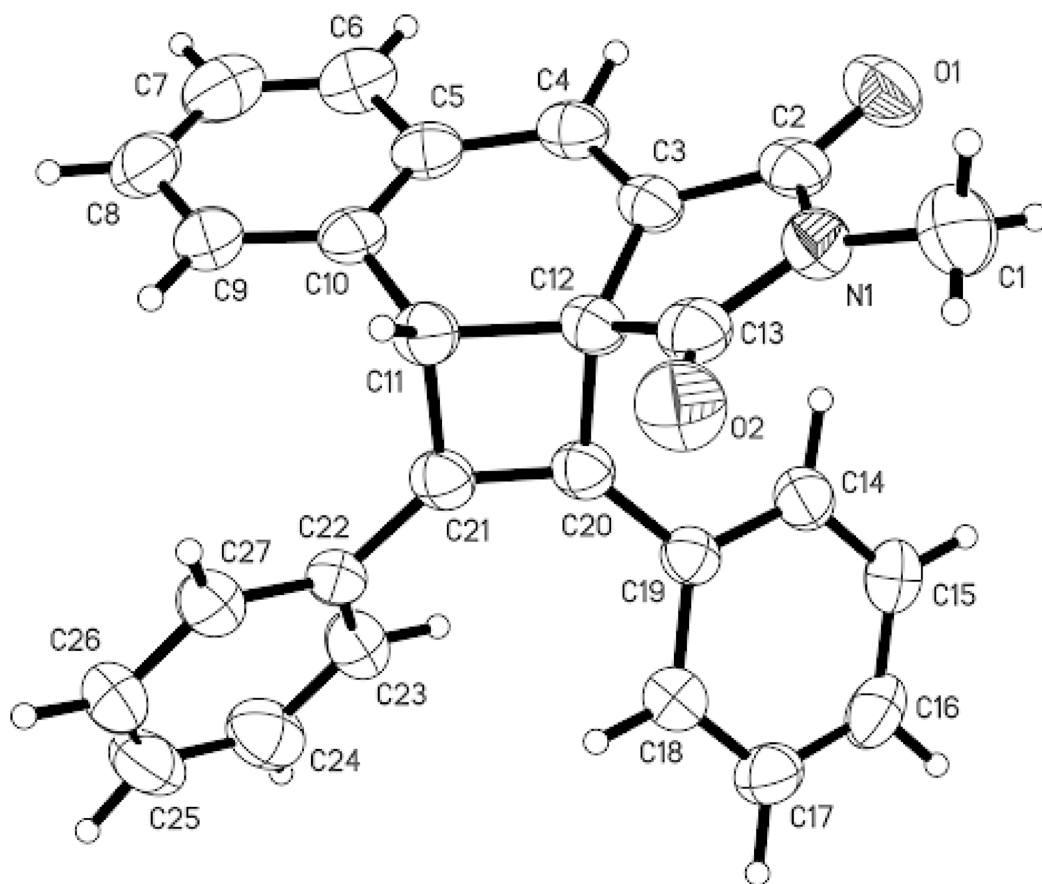


Fig. 2

